

On microstructure and room-/high-temperature properties of an $\text{Al}_2\text{O}_3/\text{Al-Cu-Mn}$ composite

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Abstract: An $\text{Al}_2\text{O}_3/\text{Al-Cu-Mn}$ composite was fabricated using a combination of ball milling and liquid-solid reaction, with a nominal composition of $\text{Al-4Cu-0.5Mn-2.8}\gamma\text{-Al}_2\text{O}_3$. The composite contains reinforcement particles, including nano-sized θ' and $\text{T}(\text{Al}_{20}\text{Cu}_2\text{Mn}_3)$ particles after T6 heat treatment, as well as in-situ synthesized nano-sized $\gamma\text{-Al}_2\text{O}_3$ particles. Tensile tests of the $\text{Al-4Cu-0.5Mn-2.8}\gamma\text{-Al}_2\text{O}_3$ composite and the Al-4Cu-0.5Mn base alloy after T6 treatment were carried out at room temperature and elevated temperatures (200 °C, 300 °C, and 400 °C). Compared with the base alloy, the yield strength of the $\text{Al-4Cu-0.5Mn-2.8}\gamma\text{-Al}_2\text{O}_3$ composite after T6 treatment increases significantly from 187 MPa to 263 MPa at room temperature. Simultaneously, at elevated temperatures, the yield strength is also enhanced, with a yield strength of 52 MPa at 400 °C for this composite. The in-situ fabricated $\gamma\text{-Al}_2\text{O}_3$ particles, mainly distributed along the grain boundaries, are supposed to play the main strengthening role, especially at high temperatures. This work acts as a reference for designing composites for high-temperature applications.

Keywords: Al matrix composite; mechanical strength; microstructure; Al_2O_3

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1 Introduction

Al-Cu-Mn-based alloys are widely used in aerospace and automotive industries due to their lightweight and high strength^[1, 2]. In recent years, extensive research has focused on utilizing elements such as Zr, Sc, Er, and Gd to further improve the tensile strength of the alloys^[3, 4]. However, the concept by increasing alloying element content to enhance the tensile properties of Al-Cu-Mn-based alloys is limited, as the alloys are highly susceptible to casting defects during solidification when the element content is high to some extent, e.g., porosity, hot tearing, and segregation, which can significantly degrade the mechanical properties of the alloys^[5]. Except for traditional alloying theory, the development of aluminum matrix composites (AMCs) has attracted much attention due to their typically low

density, high strength, and excellent high-temperature (HT) properties^[6, 7]. The desired material is achieved by controlling the type, morphology characteristics, size, and distribution of the reinforcing particles in the matrix.

Currently, AMCs are mainly fabricated by either ex-situ or in-situ methods. The ex-situ methods, such as the casting method (e.g., melt-stirring^[8] and squeeze casting^[9]), are commonly used to prepare the AMCs. However, the problem of poor wettability at the interface between the Al matrix and reinforcing particles, results in a susceptibility of the interface to contamination^[10, 11]. In contrast, in-situ methods exhibit obvious advantages in overcoming these drawbacks^[12]. The in-situ methods are mainly based on the in-situ synthesis of ceramic particles and related intermetallic compounds as reinforcements through aluminum thermal substitution reaction between Al and precursors^[13]. Reinforcements generated via in-situ methods are characterized by their non-pollution interfaces with the matrix and excellent high-temperature stability^[14]. Thus, in-situ methods have received extensive attention and development in recent years, with self-propagating synthesis^[15] and liquid-solid

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reaction^[16, 17] being commonly utilized techniques. For instance, Hu et al.^[18] prepared a novel (8.5wt.% Al_3BC +14.5wt.% Al_2O_3)/Al composite by liquid-solid reaction method, exhibiting an ultimate tensile strength (UTS) of 188 MPa and an elongation (EL) of 4.4% at 350 °C. At present, ceramic reinforcement particles are widely studied in AMCs such as TiC ^[19], SiC ^[20], Al_2O_3 ^[21], AlN ^[22], etc. Among them, Al_2O_3 particles have received more attention due to their low density, high melting point, and high strength. Nano- Al_2O_3 particles generated by in-situ methods can be distributed along grain boundaries (GBs), effectively blocking GB sliding^[23], e.g., an Al_2O_3 /Al-Cu composite was fabricated in previous work, which exhibited a UTS of 420 MPa and an EL of 5.7%^[24]. There are many metal oxides (MOs) commonly used for the preparation of Al_2O_3 -reinforced AMCs, e.g., Fe_2O_3 ^[25], CuO ^[26], Ti_2O ^[27], and ZrO_2 ^[28]. Among amounts of Al-MO systems, CuO becomes one of the most applied precursors. For instance, Rong et al.^[29] fabricated Al-5wt.% CuO, which can achieve UTS of 481 MPa, along with an EL of 16.8% at room temperature (RT). The reduced Cu atoms from CuO can be effectively dissolved into the Al matrix and subsequently precipitate as nano-sized θ' particles after aging treatment, which serve as reinforcements^[30, 31].

In this study, based on our current research^[24, 32], Al-5CuO-0.8MnO₂ (all components in this paper are wt.% unless otherwise noted) composite were fabricated via a combination of ball milling, liquid-solid reaction, and hot extrusion, aiming to design a composite with enhanced tensile properties. For comparison, an Al-4%Cu-0.5%Mn alloy was also prepared. The microstructure and mechanical properties of the composite and alloy were analyzed.

2 Experimental methods

2.1 Materials and composite fabrication

Raw materials used in this study included pure Al powder (99.9%, ~1 μm), copper oxide (CuO) powder (99.9%, ~50 nm), and manganese dioxide (MnO_2) powder (99.9%, ~50 nm). The weight ratio of Al, CuO, and MnO_2 powder was 94.2:5:0.8 and then ball milled using a planetary ball mill at 360 $\text{r}\cdot\text{min}^{-1}$ for 3 h. The ball-milled powder was put into a cylindrical rubber mold and then pressed through a cold isostatic press (LDJ200/500-380YS) at a pressure of 210 MPa to form a cylindrical blank. Then, the cylindrical blank was sintered in

a vacuum sintering furnace (ZM-39-13) at 690 °C for 15 min. The sintered billet underwent hot extrusion at 460 °C with an extrusion ratio of 20:1. Through the in-situ chemical reactions between Al, CuO, and MnO_2 , the final composite achieved was Al-4Cu-0.5Mn-2.8 γ - Al_2O_3 , namely Al-5CuO-0.8MnO₂ composite. For comparison, an Al-4%Cu-0.5%Mn alloy with the composition of Al-4Cu-0.5Mn was fabricated by hot extrusion an as-cast ingot using similar parameters^[33]. Both bars were then subjected to T6 heat treatment, which involved solid solution treatment at 520 °C for 24 h followed by immediately quenched in water and an aging treatment at 185 °C for 48 h with subsequent air cooling.

2.2 Microstructure characterization

Phase characterization of Al-5CuO-0.8MnO₂ composite and Al-4Cu-0.5Mn alloy was carried out using X-ray diffraction (XRD, Rigaku D/max-rB, Tokyo, Japan) with Cu K α radiation. The microstructure was analyzed using a field emission scanning electron microscope (FESEM, SU-70, Hitachi, Tokyo, Japan) equipped with an energy dispersive X-ray spectrometer (EDS). Grain size and Kernel average misorientation (KAM) were analyzed using a FESEM (FEI Verios 5 UC, FEI Company, Hillsboro, OR, USA) equipped with an EBSD detector (Symmetry S2, UK), and the data were processed using Channel 5 software. Transmission electron microscopy (TEM, FEI Talos F200X, 200 kV) with EDS capability was employed to observe the morphology and distribution of nano-precipitated particles. TEM samples were initially mechanically polished to a thickness of approximately 50 μm , then shaped into 3 mm circular discs using a perforating machine, and finally polished to a central perforation using an ion polishing system (Gatan 695, Pleasanton, CA, USA) for detailed observation.

2.3 Tensile test

The T6 heat-treated Al-5CuO-0.8MnO₂ composite and Al-4Cu-0.5Mn alloy test bars were machined into “dog-bone” shaped specimens for RT and HTs (200 °C, 300 °C, and 400 °C) tensile testing. The RT and HT tensile tests were carried out on a universal tensile testing machine (CMT700) with a loading rate of 2 $\text{mm}\cdot\text{min}^{-1}$. To ensure the accuracy of the tests, three specimens were tested at each temperature and the average value was taken. The required specimen sizes of RT and HT tensile tests are shown in Fig. 1.

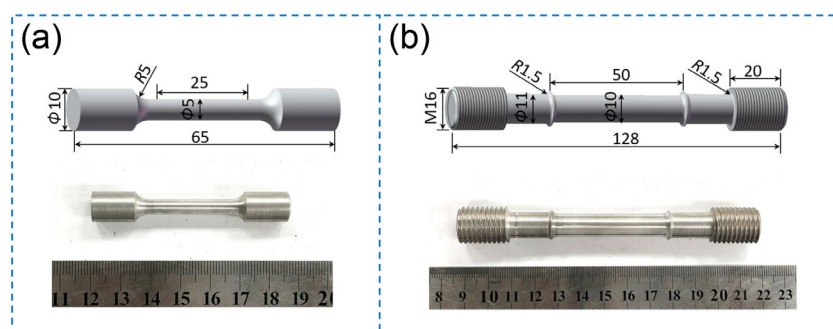


Fig. 1: Dimensions of tensile test specimens at different temperatures: (a) RT; (b) HT (unit: mm)

3 Results and discussion

Figure 2 shows the microstructure of the Al-5CuO-0.8MnO₂ composite and the Al-4Cu-0.5Mn alloy along the longitudinal section. The Al₂Cu phases determined by EDS analysis [(Fig. 2(b)] in the as-extruded composite are distributed along the extrusion direction [Fig. 2(a)]. It can also be seen that the quantity of Al₂Cu phases is significantly decreased in Fig. 2(c), indicating that the Al₂Cu phases have been dissolved into the Al matrix during the solution treatment. Figure 2(d) shows a magnified area of the marked Region B in Fig. 2(c), revealing that Al and CuO have synthesized a large number

of Al₂O₃ particles by an in-situ reaction. Figure 2(e) shows the microstructure of the aging-treated composite. Many fine rod-like white particles (indicated by yellow arrows) and larger gray particles (indicated by a blue arrow) are found in the yellow magnified area, and further analysis will be conducted using TEM. Figures 2(f-g) exhibit the microstructure of the as-extruded and solid solution-treated Al-4Cu-0.5Mn base alloy. The distribution of the Al₂Cu phase confirmed by mapping [Fig. 2(h)] is also oriented along the extrusion direction in Fig. 2(f). In addition, the Al₂Cu phase during the solution treatment [Fig. 2(g)] is also largely incorporated into the Al matrix compared to the as-extruded base alloy [Fig. 2(f)].

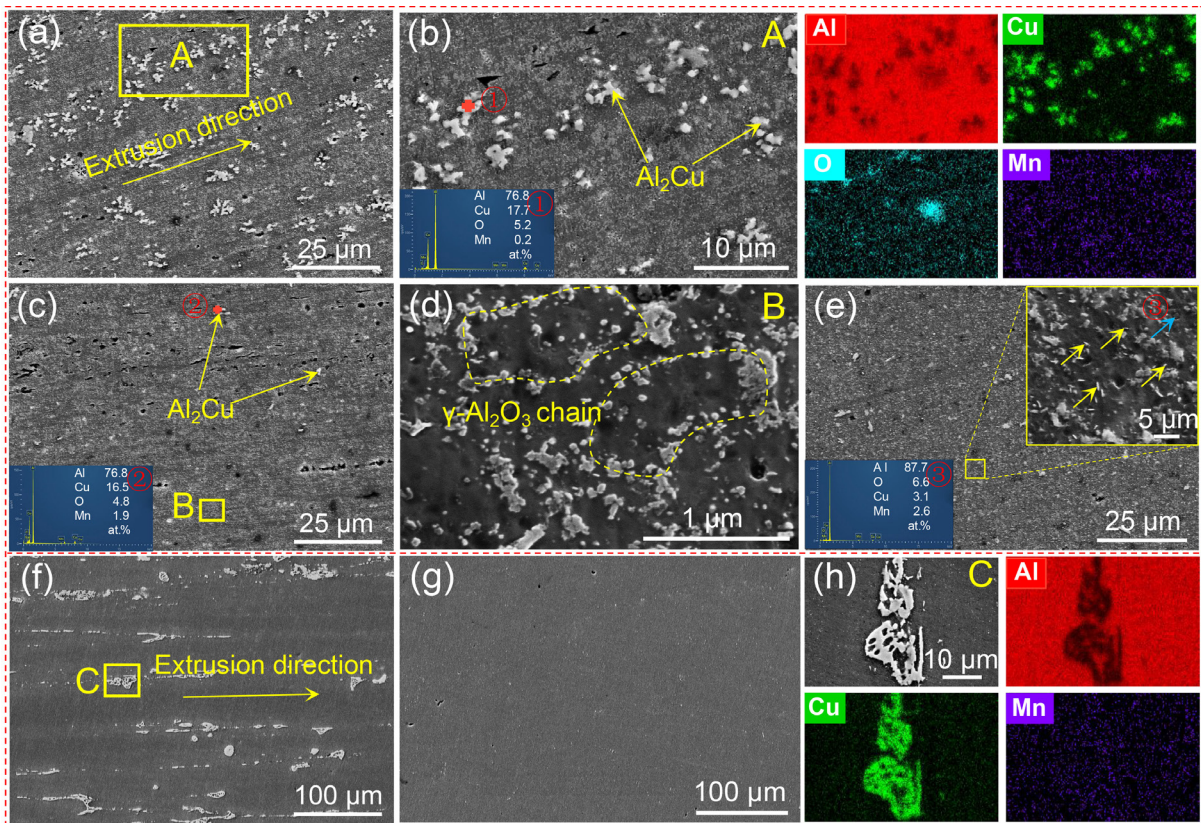


Fig. 2: SEM microstructures of Al-5CuO-0.8MnO₂ composite (a-e) and Al-4Cu-0.5Mn alloy (f-h) in longitudinal section: (a) microstructure of as-extruded composite; (b) magnified area of marked Region A in (a) and its mapping (the inset shows point analysis at Point 1); (c) microstructure of solution-treated composite (the inset shows the point analysis at Point 2); (d) the magnified area of marked Region B in (c); (e) microstructure of aging-treated composite (the inset shows point analysis at Point 3); (f, g) microstructure of as-extruded and solution-treated base alloy, respectively; (h) mapping of magnified area of marked Region C in (f)

Figure 3 exhibits the XRD patterns of the Al-5CuO-0.8MnO₂ composite with the as-extruded, solid solution-treated, and aging-treated methods along the longitudinal section. Based on the standard XRD database^[34], the Al₂Cu peak (PDF#89-1981) and γ -Al₂O₃ peak (PDF#23-1009) are successfully detected, which is consistent with our previous work^[32]. It is also found that the diffraction peaks of Al₂Cu disappear in the solution treatment and reappear in the aging treatment. This indicates the dissolution of the Al₂Cu phases and reprecipitation of the nano-sized θ' (Al₂Cu) phases, respectively, aligning with the observations presented in Fig. 2.

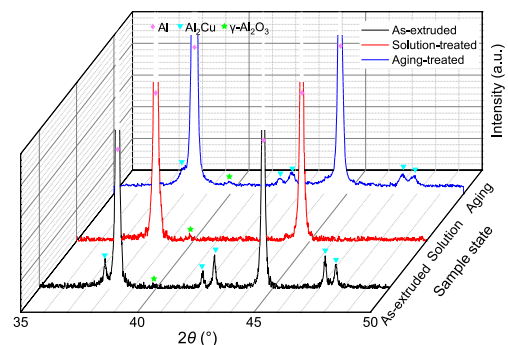


Fig. 3: XRD patterns of Al-5CuO-0.8MnO₂ composite after extrusion, solution, and aging

Typical EBSD results of the Al-5CuO-0.8MnO₂ composite and the Al-4Cu-0.5Mn alloy after T6 treatment are shown in Fig. 4. As can be found in Figs. 4(a) and (b), the equiaxed grains and elongated grains together constitute the microstructure of the Al-5CuO-0.8MnO₂ composite, while the Al-4Cu-0.5Mn alloy is mainly composed of elongated grains. The statistical grain size of the Al matrix in the Al-5CuO-0.8MnO₂ composite is 0.71 μm , which is much smaller than that of 82.95 μm in the Al-4Cu-0.5Mn alloy, as shown in Figs. 4(c) and (d). Figures 4(e) and (f) demonstrate the KAM map of the Al-5CuO-0.8MnO₂ composite and the Al-4Cu-0.5Mn alloy after T6 treatment, respectively. The Al-5CuO-0.8MnO₂ composite has a larger KAM value than the Al-4Cu-0.5Mn alloy [seen in Fig. 4(g)], indicating a higher dislocation density in the composite [35]. Based on Figs. 4(e) and (f), the density of geometrically necessary dislocations (GNDs) is calculated to be $6.2 \times 10^{13} \text{ m}^{-2}$ for the Al-5CuO-0.8MnO₂ composite and $5.4 \times 10^{11} \text{ m}^{-2}$ for the Al-4Cu-0.5Mn alloy, respectively. This phenomenon, combined with the fine matrix grain size, may be due to the formation of nano-sized Al₂O₃ particles, which can cause an increase in residual stresses or micro-zone deformation inhomogeneity in the composite. Figure 4(h) displays the statistics of the Schmid factor for the Al-5CuO-0.8MnO₂ composite and the Al-4Cu-0.5Mn alloy after T6 treatment, i.e., 0.43 and 0.47, respectively. This suggests that the generation of nano-sized Al₂O₃ particles does not affect the deformability of the composite [36].

Figure 5 presents the typical microstructure of the Al-5CuO-0.8MnO₂ composite after T6 treatment. The presence of needle-like, plate-like, and block-like particles is observed in Fig. 5(a). The EDS mapping in Figs. 5(a1–a3) as well as the inserted selected area electron diffraction (SAED) pattern confirms that these phases are Al₂Cu. Figure 5(b) shows a magnified image of the marked Region A in Fig. 5(a). Both the needle and plate phases are identified as θ' (Al₂Cu) by high-resolution analysis of the marked Regions B and C, respectively [Figs. 5(c) and (d)]. Two kinds of morphologies

occur since they are viewed from different directions on the flake-like shape. Meanwhile, the inserted D and E fast fourier transform (FFT) images further confirm this result. The diameter of θ' phase was counted in Fig. 5(e). The average length of the θ' phase in the Al-5CuO-0.8MnO₂ composite reaches 103 nm. Figures 5(f) and (g) exhibit the high-angle annular dark field (HAADF) and bright field (BF) images of the Al-4Cu-0.5Mn alloy after T6 treatment, respectively, and the corresponding SAED plots are shown in Fig. 5(h). It can be found that there are both needle-like and plate-like θ' particles. The average diameter of θ' phases in Al-4Cu-0.5Mn alloy is 131 nm. Meanwhile, it can be clearly observed that the particle density of θ' in Al-5CuO-0.8MnO₂ composite [Fig. 5(a)] is much less than that in Al-4Cu-0.5Mn alloy [Fig. 5(f)].

Figures 6(a) and (b) illustrate the HAADF images of both T6-treated Al-5CuO-0.8MnO₂ composite and Al-4Cu-0.5Mn alloy as well as their corresponding elemental distributions [Figs. 6(a1) and (b1)]. It can be clearly seen that Mn-rich phases with block-like and rod-like morphologies are present in both materials. Quantitative analysis indicates that the density of Mn-rich phases in the Al-5CuO-0.8MnO₂ composite is less than that in the Al-4Cu-0.5Mn alloy. It has been reported that another reinforcing phase, the T(Al₂₀Cu₂Mn₃) phase, exists in the Al-Cu-Mn-based alloys [1]. The block-like T phase [Fig. 2(e)] is probably obtained from the decomposition of Phase I (Al₁₃Cu₄Mn₃) during the solid solution treatment, while the rod-like phase forms by direct precipitation during aging treatment [1, 37]. This corresponds to the phases identified by the blue and yellow arrows in Fig. 2(e). Figure 6(c) shows a TEM image of a single Al grain and the EDS mappings [Fig. 6(c1)]. It is observed that γ -Al₂O₃ is mainly distributed along the GBs of Al grain, with only a few particles presented within the grain itself.

Engineering stress-strain curves for the Al-5CuO-0.8MnO₂ composite and Al-4Cu-0.5Mn alloy after T6 treatment were tested at RT, 200 °C, 300 °C, and 400 °C, as depicted in Fig. 7. The corresponding yield strength (YS), UTS, and EL

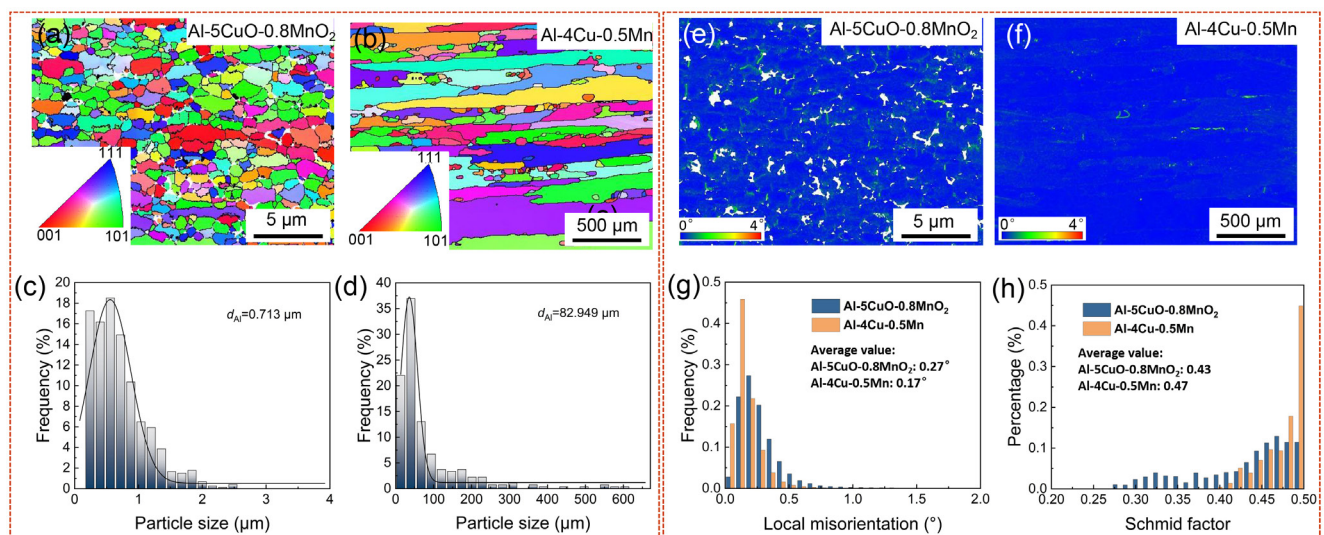


Fig. 4: EBSD characterization of Al-5CuO-0.8MnO₂ composite and Al-4Cu-0.5Mn alloy after T6 treatment: (a, b) orientation map; (c, d) grain size; (e, f) KAM map; (g) KAM distribution; (h) Schmid factor distribution

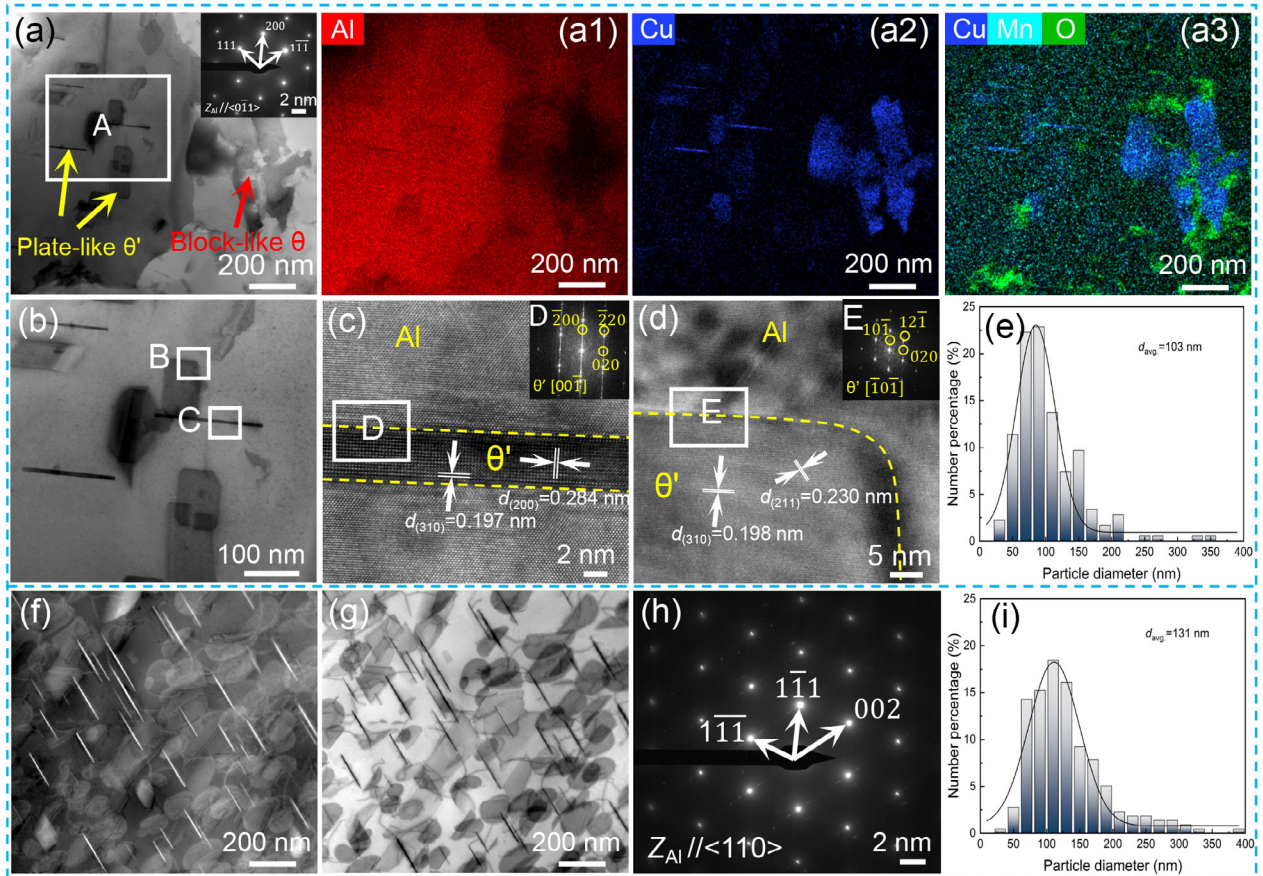


Fig. 5: TEM microstructures and diameter size distribution of Al-5CuO-0.8MnO₂ composite (a-e) and Al-4Cu-0.5Mn alloy (f-i) after T6 treatment: (a) BF image and SAED pattern; (a1-a3) EDS mapping of Al-5CuO-0.8MnO₂ composite; (b) magnified area of marked Region A in (a); (c) and (d) are the high-resolution of magnified area of marked Region B and C in (c) (inserted FFT D and FFT E recorded the marked Regions D and E, respectively); (e) diameter size distribution of θ' particles in Al-5CuO-0.8MnO₂ composite; (f and g) HAADF and BF images of Al-4Cu-0.5Mn alloy; (h) SAED pattern of (f); (i) diameter size distribution of θ' particles in Al-4Cu-0.5Mn alloy

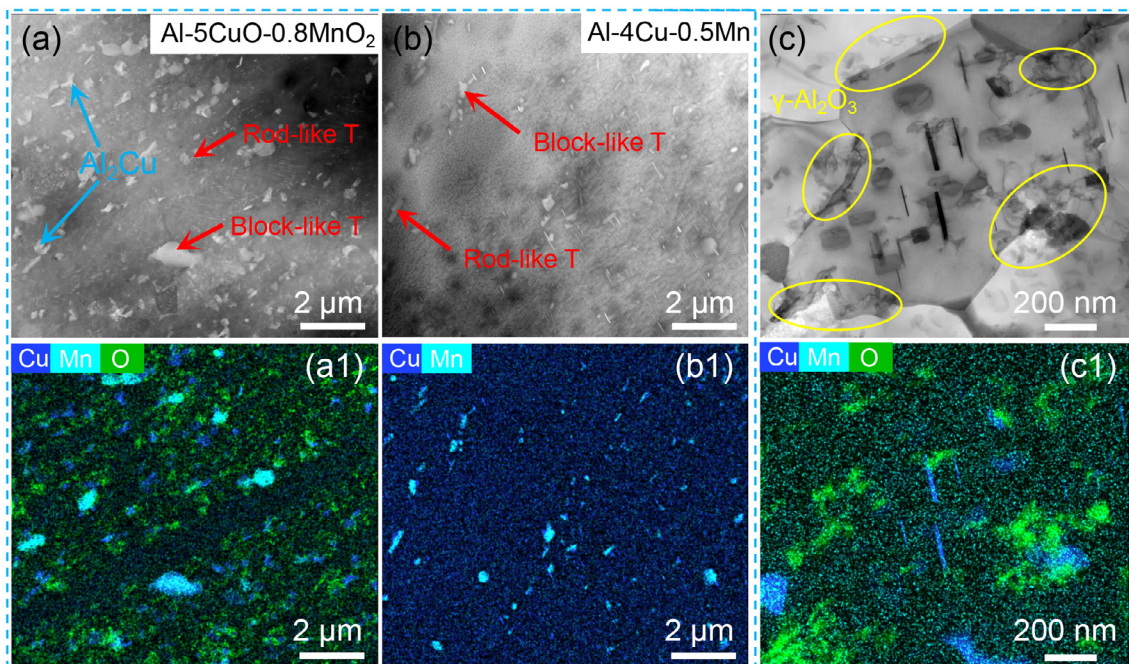


Fig. 6: TEM microstructure of Al-5CuO-0.8MnO₂ composite and Al-4Cu-0.5Mn alloy after T6 treatment: HAADF images (a, b) and EDS mapping (a1, b1) of Al-5CuO-0.8MnO₂ composite and Al-4Cu-0.5Mn alloy, respectively; HAADF image (c) and EDS mapping (c1) of a single grain in Al-5CuO-0.8MnO₂ composite

results are exhibited in Table 1. It is found that the YS and UTS values of both the Al-5CuO-0.8MnO₂ composite and Al-4Cu-0.5Mn alloy after T6 treatment decrease with increasing testing temperature, indicating that the Al matrix undergoes high-temperature softening. At RT, the YS of the Al-5CuO-0.8MnO₂ composite after T6 treatment is 263 MPa, which is a 41% increase compared to 187 MPa for the Al-4Cu-0.5Mn alloy after T6 treatment. Moreover, the YS of T6-treated Al-5CuO-0.8MnO₂ composite tested at 200 °C, 300 °C, and 400 °C are 171 MPa, 98 MPa, and 52 MPa, respectively, which are 14%, 8%, and 53% higher than that of T6-treated Al-4Cu-0.5Mn alloy. This indicates that the Al-5CuO-0.8MnO₂ composite treated by T6 exhibits superior high-temperature strength than that of the Al-4Cu-0.5Mn alloy treated by T6, probably related to the presence of γ -Al₂O₃ particles.

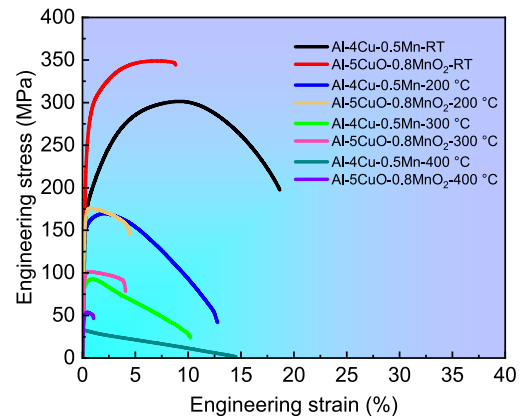


Fig. 7: Typical tensile engineering stress-strain curves of Al-5CuO-0.8MnO₂ composite and Al-4Cu-0.5Mn alloy after T6 treatment tested at RT, 200 °C, 300 °C, and 400 °C

Table 1: Mechanical properties of Al-5CuO-0.8MnO₂ composite and Al-4Cu-0.5Mn alloy after T6 treatment

Tensile test temperatures	Materials	YS (MPa)	UTS (MPa)	EL (%)
RT	Al-4Cu-0.5Mn	187±18	305±15	18.3±0.4
	Al-5CuO-0.8MnO ₂	263±3	347±7	9.0±0.6
200 °C	Al-4Cu-0.5Mn	150±19	169±15	12.8±1.7
	Al-5CuO-0.8MnO ₂	171±1	176±1	5.1±0.6
300 °C	Al-4Cu-0.5Mn	91±3	96±3	9.9±0.3
	Al-5CuO-0.8MnO ₂	98±1	100±1	4.1±0.1
400 °C	Al-4Cu-0.5Mn	34±1	34±1	14.9±0.4
	Al-5CuO-0.8MnO ₂	52±2	53±2	1.1±0.1

The fracture microstructures of the Al-5CuO-0.8MnO₂ composite and the Al-4Cu-0.5Mn alloy after T6 treatment through tensile testing at RT, 200 °C, 300 °C, and 400 °C are shown in Fig. 8. Tearing ridges and various sizes of dimples are observed on the Al-4Cu-0.5Mn alloy fracture surfaces in Figs. 8(a–d), indicating typical ductile fracture characteristics. The presence of large and deep dimples at room-temperature and high-temperature suggests that this alloy has good plasticity. In the Al-5CuO-0.8MnO₂ composite [Figs. 8(e) and (f)], similar ductile fracture features are observed, with dimples of different sizes. However, compared to the Al-4Cu-0.5Mn alloy, these dimples are smaller and shallower in size, suggesting lower deformability for the Al-5CuO-0.8MnO₂ composite [38]. In addition, the dimple features are hardly visible in the fracture surface at 300 °C and 400 °C, as shown in Figs. 8(g) and (h), where intergranular fracture characteristics predominate. It has been reported that the dynamic recovery at GBs may lead to localized deformation, potentially initiating the generation of voids and cracks [39, 40]. As displayed in Figs. 8(g), relative sliding (indicated by yellow arrows) is found between neighboring grains on the fracture surfaces, indicating the

occurrence of dynamic recovery at GBs. As a result, voids (red arrows) generate, which in turn leads to the fracture of the composite.

To further clarify the changes in strength and ductility of the T6-treated Al-5CuO-0.8MnO₂ composite tensile tested at room-temperature and high-temperature, the strengthening mechanism and tensile fracture morphologies of the Al-5CuO-0.8MnO₂ composite (Fig. 8) were analyzed. Despite the lower quantitative densities of θ' and T(Al₂₀Cu₂Mn₃) reinforcements in the Al-5CuO-0.8MnO₂ composite compared to the Al-4Cu-0.5Mn alloy, as can be easily found in Figs. 5(b) and (f), the YS of the Al-5CuO-0.8MnO₂ composite still surpasses that of the Al-4Cu-0.5Mn alloy at room temperature. This enhancement is attributed to grain refinement strengthening and load-transfer strengthening due to the presence of nano-sized Al₂O₃ particles. The Hall-Petch relationship indicates that finer grains have higher YS [41]. After T6 treatment, the average grain size of the Al-5CuO-0.8MnO₂ composite is much smaller compared to that of the Al-4Cu-0.5Mn alloy (Fig. 4), resulting in the increased YS of the Al-5CuO-0.8MnO₂ composite. In addition, the in-situ synthesized nano-sized Al₂O₃

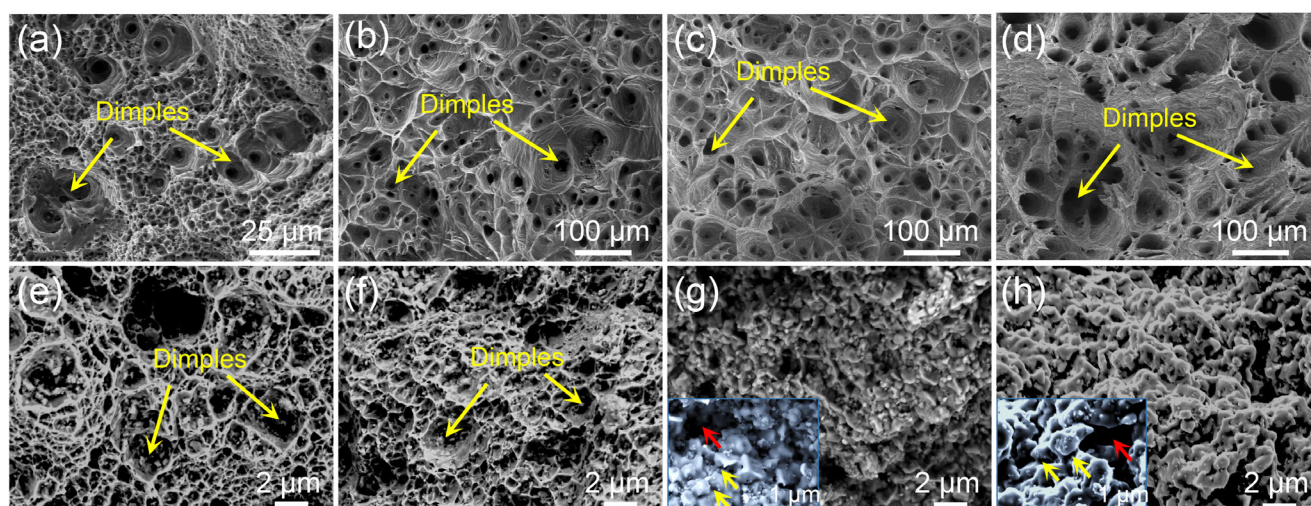


Fig. 8: Tensile fracture morphologies of Al-4Cu-0.5Mn alloy (a-d) and Al-5CuO-0.8MnO₂ composite (e-h) after T6 treatment: (a, e) RT; (b, f) 200 °C; (c, g) 300 °C; (d, h) 400 °C

particles provide additional strengthening effects. In general, their strengthening effects are usually realized in two ways: direct strengthening via load transfer and indirect secondary strengthening resulting from dislocation accumulation, which includes the Orowan strengthening mechanism and the strengthening due to GNDs^[42-44]. For instance, the detected GNDs density of the Al-5CuO-0.8MnO₂ composite is far more larger than that of the Al-4Cu-0.5Mn alloy, indicating the existence of the abovementioned strengthening mechanism. Therefore, the combined effects of multiple strengthening mechanisms result in a favorable YS of the Al-5CuO-0.8MnO₂ composite at room temperature.

The θ' phase and T(Al₂₀Cu₂Mn₃) phase can play a certain role in strengthening at lower temperatures but have limited effectiveness at higher temperatures, such as 300 °C and 400 °C, as discussed in this study. In comparison, the in-situ generated nano-sized Al₂O₃ particles maintain relatively good stability at high temperatures. Therefore, it is inferred that γ -Al₂O₃ particles play a major role in the strengthening of the Al-5CuO-0.8MnO₂ composite at high temperatures. Sliding of GBs and rotation of crystals at high temperatures lead to the release of dislocation accumulation at the GBs, thereby weakening their reinforcement capacity and softening the Al matrix^[45]. However, the pinning effect of fine particles or nanoparticles can effectively prevent GB movement, thus improving the high temperature performance of the composites^[39, 45, 46]. As illustrated in Fig. 6(c), the γ -Al₂O₃ particles are agglomerated along the GBs, exerting a pinning effect that contributes to improved high temperature tensile strength of the Al-5CuO-0.8MnO₂ composite.

Compared Fig. 8(f) with Fig. 8(e), it can be found that the number and size of dimples on the fracture surface of the Al-5CuO-0.8MnO₂ composite decrease to some extent when the testing temperature increases to 200 °C, resulting in lower deformability of the composite^[38]. In addition, the dislocation density decreases at high temperatures due to the onset of dynamic recovery, leading to a reduction in the

work-hardening effect^[38]. Meanwhile, due to the occurrence of dynamic recovery, the dislocations interact with the Al₂O₃ particles at the GBs, leading to localized plastic deformation and the formation of cavities^[40]. This interaction ultimately results in the development of intergranular fracture [Figs. 8(g) and (h)]. Collectively, the combined effect contributes to the decrease in the elongation of the Al-5CuO-0.8MnO₂ composite with increasing temperatures (Fig. 7).

4 Conclusions

In this work, the Al-5CuO-0.8MnO₂ composite was fabricated with a nominal composition of Al-4Cu-0.5Mn-2.8 γ -Al₂O₃. The effects of intermetallic compounds and γ -Al₂O₃ particles on microstructure evolution, mechanical properties, and strengthening mechanisms at RT, 200 °C, 300 °C, and 400 °C were systematically investigated. An Al-4Cu-0.5Mn alloy was fabricated for comparison. The key results are summarized below:

(1) With a combination of sintering, hot extrusion, and T6 heat treatment, the achieved Al-5CuO-0.8MnO₂ composite contains θ' , T(Al₂₀Cu₂Mn₃) phases, and γ -Al₂O₃ particles. The γ -Al₂O₃ particles are mainly distributed along the GBs, resulting in a significantly finer matrix grain size in the T6 treated Al-5CuO-0.8MnO₂ composite compared to the T6 treated Al-4Cu-0.5Mn alloy.

(2) Compared with the T6-treated Al-4Cu-0.5Mn base alloy, the YS of the T6 treated Al-5CuO-0.8MnO₂ composite is enhanced from 187 MPa to 263 MPa at room temperatures, 150 MPa to 171 MPa at 200 °C, 91 MPa to 98 MPa at 300 °C, and 34 MPa to 52 MPa at 400 °C, respectively. The improvement is attributed to the positive strengthening effects of the in-situ γ -Al₂O₃ nanoparticles.

(3) By promoting chemical reactions between the Al matrix and metal oxides, Al₂O₃-reinforced AMCs may overcome the weakness of traditional Al alloys prepared using alloying methods, particularly in high temperature applications.

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Conflict of interest

The authors declare that they have no competing financial interests or personal relationships that could have appeared to influence the work in this paper.

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